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CHAPTER 6 REVIEW Chemical Bonding  
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bond approach each other, the

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when atoms share one or more pairs of electrons. the region of high probability that is occupied by an individu.... a molecular orbital.

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## Chapter 6 Covalent Bonds

Holt Chemistry - Chapter 6 Vocabulary. the region of high probability that is occupied by an individual electron as it travels with a wavelike motion in the three dimensional space around one of two or more associated nuclei. This activity was created by a Quia Web subscriber.

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Chapter 6 Vocabulary. Description. Modern Chemistry : Holt, Rinehart, and Winston. Total Cards. 28. Subject. Chemistry. Level. 11th Grade. Created. 03/23/2011. [Click here to study/print these flashcards.](#) ... A covalent bond in which the bonding electrons are shared equally by the bonded atoms, resulting in a balanced distribution of electrical ...

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## Chapter 6 Covalent Bonds

chemical bonding.

### **Holt McDougal Modern Chemistry Chapter 6: Chemical Bonding ...**

Holt Modern Chemistry Review CHAPTER 6: CHEMICAL BONDING The following pages contain the bulk (but not all) of the information for the chapter 6 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test. \*\*\*\*\* Test items will be recall, examples, and/or ...

### **CHEMISTRY CHAPTER 6 REVIEW.docx - Holt Modern Chemistry ...**

\_\_\_\_\_ are formed by covalent bonding and are represented by molecular formulas. Molecules: Formula units are formed by \_\_\_\_\_ bonding and are represented by empirical formulas. ionic: Nonpolar Covalent Bond: a bond in which 2 like metallic atoms share a pair

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of electrons equally: Polar Covalent Bond: a bond in which 2 different atoms share ...

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Polarity of Molecules Naming  
Compounds COVALENT BONDING  
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Bonds continued PROPERTIES OF  
MOLECULES Remember that an atom is  
the smallest piece of an element that  
still has the properties of that element.  
In the same way, a molecule is the  
smallest piece of a covalent compound  
that has the properties of that  
compound.

## **Holt Chemistry Bonds Compounds Concept Review Answers**

Most organic chemicals are covalent  
compounds, which is why we introduce  
organic chemistry here. By convention,  
compounds containing carbonate ions  
and bicarbonate ions, as well as carbon  
dioxide and carbon monoxide, are not

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considered part of organic chemistry,  
even though they contain carbon.

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